

SURENDRANATH COLLEGE

DEPARTMENT OF CHEMISTRY, DR.HSB CLASSES

QUANTUM MECHANICS 5TH SEM

Angular momentum: Commutators rules, quantization of square of total angular momentum and z-component; Rigid rotator model of rotation of diatomic molecules; Schrodinger equation, transformation of spherical coordinates; separation of variables. Spherical harmonics; discussion of solution

Note: \hbar (h-bar) Iran N. Levine, Quantum Chemistry, 7ed, P – 9

Angular Momentum

Consider a particle of mass “m” revolving around at a fixed point. The angular momentum

$$L = r \times mu = r \times p$$

The cross product $r \times p$ is the vector L perpendicular to the plane formed by the vectors r and p .

Where “r” is the vector from the fixed point to mass point

p is the linear momentum vector

In classical mechanics, component of linear momentum vector are

$$L_x = yp_z - zp_y \quad L_y = zp_x - xp_z \quad L_z = xp_y - yp_x$$

$$L^2 = LL = L_x^2 + L_y^2 + L_z^2$$

Note: $i \times i = j \times j = k \times k = 0$

$$i \times j = k, \quad j \times k = i, \quad k \times i = j$$

$$j \times i = -k, \quad k \times j = -i, \quad i \times k = -j$$

$$i \cdot i = j \cdot j = k \cdot k = 1$$

$$i \cdot j = j \cdot k = k \cdot i = 0$$

Corresponding to the quantum mechanical operator $\hat{p}_x = -i\hbar \frac{\partial}{\partial x}$ $\hat{p}_y = -i\hbar \frac{\partial}{\partial y}$ $\hat{p}_z = -i\hbar \frac{\partial}{\partial z}$

$$\hat{L}_x = -i\hbar \left(y \frac{\partial}{\partial z} - z \frac{\partial}{\partial y} \right) \quad \hat{L}_y = -i\hbar \left(z \frac{\partial}{\partial x} - x \frac{\partial}{\partial z} \right) \quad \hat{L}_z = -i\hbar \left(x \frac{\partial}{\partial y} - y \frac{\partial}{\partial x} \right)$$

$$\hat{L}^2 = |\hat{L}|^2 = \hat{L}L = \hat{L}_x^2 + \hat{L}_y^2 + \hat{L}_z^2$$

$$[\hat{L}_x, \hat{L}_y] = i\hbar \hat{L}_z \quad [\hat{L}_y, \hat{L}_z] = i\hbar \hat{L}_x \quad [\hat{L}_z, \hat{L}_x] = i\hbar \hat{L}_y \quad [\hat{L}^2, \hat{L}_z] = 0$$

$$[\hat{L}_z, \hat{L}_x] = -[\hat{L}_x, \hat{L}_z]$$

Prove

$$[\hat{L}_x, \hat{L}_y] = i\hbar \hat{L}_z$$

$$[\hat{L}_x, \hat{L}_y] = \hat{L}_x \hat{L}_y - \hat{L}_y \hat{L}_x$$

$$\begin{aligned} \text{Now } \hat{L}_x \hat{L}_y &= \left\{ -i\hbar \left(y \frac{\partial}{\partial z} - z \frac{\partial}{\partial y} \right) \right\} \left\{ -i\hbar \left(z \frac{\partial}{\partial x} - x \frac{\partial}{\partial z} \right) \right\} \\ &= -\hbar^2 \left(y \frac{\partial}{\partial z} z \frac{\partial}{\partial x} - y \frac{\partial}{\partial z} x \frac{\partial}{\partial z} - z \frac{\partial}{\partial y} z \frac{\partial}{\partial x} + z \frac{\partial}{\partial y} x \frac{\partial}{\partial z} \right) \\ &= -\hbar^2 \left(y \frac{\partial}{\partial x} + yz \frac{\partial^2}{\partial z \partial x} - yx \frac{\partial^2}{\partial z^2} - z^2 \frac{\partial^2}{\partial y \partial x} + zx \frac{\partial^2}{\partial y \partial z} \right) \end{aligned}$$

$$\begin{aligned}
\text{Similarly } \hat{L}_y \hat{L}_x &= \left\{ -i\hbar \left(z \frac{\partial}{\partial x} - x \frac{\partial}{\partial z} \right) \right\} \left\{ -i\hbar \left(y \frac{\partial}{\partial z} - z \frac{\partial}{\partial y} \right) \right\} \\
&= -\hbar^2 \left(z \frac{\partial}{\partial x} y \frac{\partial}{\partial z} - z \frac{\partial}{\partial x} z \frac{\partial}{\partial y} - x \frac{\partial}{\partial z} y \frac{\partial}{\partial z} + x \frac{\partial}{\partial z} z \frac{\partial}{\partial y} \right) \\
&= -\hbar^2 \left(zy \frac{\partial^2}{\partial x \partial z} - z^2 \frac{\partial^2}{\partial x \partial y} - xy \frac{\partial^2}{\partial z^2} + x \frac{\partial}{\partial y} + xz \frac{\partial^2}{\partial z \partial y} \right)
\end{aligned}$$

$$\begin{aligned}
\text{Hence, } [\hat{L}_x, \hat{L}_y] &= \hat{L}_x \hat{L}_y - \hat{L}_y \hat{L}_x \\
&= \hbar^2 \left(x \frac{\partial}{\partial y} - y \frac{\partial}{\partial x} \right) = -i^2 \hbar^2 \left(x \frac{\partial}{\partial y} - y \frac{\partial}{\partial x} \right) = i\hbar \left\{ -i\hbar \left(x \frac{\partial}{\partial y} - y \frac{\partial}{\partial x} \right) \right\} = i\hbar \hat{L}_z
\end{aligned}$$

Note: L_x and L_y or L_y and L_z cannot be determined simultaneously and precisely, because they do not commute.

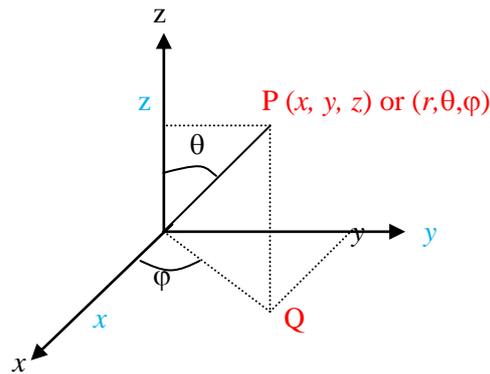
Q. L^2 and L_z can be determined simultaneously and precisely, because they commute.

$$\begin{aligned}
[\hat{L}^2, \hat{L}_z] &= [\hat{L}_x^2 + \hat{L}_y^2 + \hat{L}_z^2, \hat{L}_z] = \hat{L}_x^2 \hat{L}_z + \hat{L}_y^2 \hat{L}_z + \hat{L}_z^2 \hat{L}_z - \hat{L}_z \hat{L}_x^2 - \hat{L}_z \hat{L}_y^2 - \hat{L}_z \hat{L}_z^2 \\
&= [\hat{L}_x^2, \hat{L}_z] + [\hat{L}_y^2, \hat{L}_z] + [\hat{L}_z^2, \hat{L}_z]
\end{aligned}$$

$$\begin{aligned}
\text{Now } [\hat{L}_x^2, \hat{L}_z] &= \hat{L}_x^2 \hat{L}_z - \hat{L}_z \hat{L}_x^2 = \hat{L}_x \hat{L}_x \hat{L}_z - \hat{L}_z \hat{L}_x \hat{L}_x \\
&= \hat{L}_x \hat{L}_x \hat{L}_z - \hat{L}_x \hat{L}_z \hat{L}_x + \hat{L}_x \hat{L}_z \hat{L}_x - \hat{L}_z \hat{L}_x \hat{L}_x = \hat{L}_x (\hat{L}_x \hat{L}_z - \hat{L}_z \hat{L}_x) + (\hat{L}_x \hat{L}_z - \hat{L}_z \hat{L}_x) \hat{L}_x \\
&= \hat{L}_x [\hat{L}_x, \hat{L}_z] + [\hat{L}_x, \hat{L}_z] \hat{L}_x = -i\hbar \hat{L}_x \hat{L}_y - i\hbar \hat{L}_y \hat{L}_x = -i\hbar (\hat{L}_x \hat{L}_y + \hat{L}_y \hat{L}_x)
\end{aligned}$$

$$\text{Similarly } [\hat{L}_y^2, \hat{L}_z] = i\hbar (\hat{L}_x \hat{L}_y + \hat{L}_y \hat{L}_x) \quad \text{and} \quad [\hat{L}_z^2, \hat{L}_z] = 0$$

Transformation of Cartesian coordinates of Laplacian operator to Spherical polar coordinates



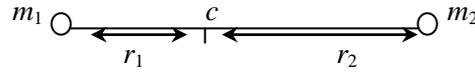
$$z = r \cos \theta, \quad x = r \sin \theta \cos \phi \quad \text{and} \quad y = r \sin \theta \sin \phi$$

$$x^2 + y^2 + z^2 = r^2$$

$$\begin{aligned}
\text{Laplacian operator } \nabla^2 &= \frac{\partial^2}{\partial x^2} + \frac{\partial^2}{\partial y^2} + \frac{\partial^2}{\partial z^2} \\
&= \frac{1}{r^2} \frac{\partial}{\partial r} \left(r^2 \frac{\partial}{\partial r} \right) + \frac{1}{r^2 \sin \theta} \frac{\partial}{\partial \theta} \left(\sin \theta \frac{\partial}{\partial \theta} \right) + \frac{1}{r^2 \sin^2 \theta} \frac{\partial^2}{\partial \phi^2}
\end{aligned}$$

Three Dimensional Rotation: A Rigid Rotator

Consider diatomic molecule with mass m_1 and m_2 separated by fixed distance r and centre of mass is c . The distance from m_1 is r_1 and distance from m_2 is r_2 i.e $r_1 + r_2 = r$.



Now, $m_1 r_1 = m_2 r_2$ So, $m_1 (r - r_2) = m_2 r_2$ or, $(m_1 + m_2) r_2 = m_1 r$

$$\therefore r_2 = \frac{m_1 r}{m_1 + m_2} \quad r_1 = \frac{m_2 r}{m_1 + m_2}$$

Moment of inertia $I = m_1 r_1^2 + m_2 r_2^2 \quad I = \frac{m_1 m_2 r^2}{m_1 + m_2} = \mu r^2$

Where, reduced mass $\mu = \frac{m_1 m_2}{m_1 + m_2}$

Now kinetic energy $T = \frac{1}{2} m_1 v_1^2 + \frac{1}{2} m_2 v_2^2 = \frac{1}{2} \omega^2 I = \frac{1}{2} \frac{L^2}{I} = \frac{1}{2} \frac{L^2}{\mu r^2}$

[$v = \omega r$, $L = \omega I$]

Quantum mechanics

Hamiltonian operator $\hat{H} = \hat{T} + \hat{V}$

For rigid rotator potential energy $V = 0$

So, $\hat{H} = \hat{T} = \frac{\hat{L}^2}{2\mu r^2} = -\frac{\hbar^2}{8\pi^2 \mu} \nabla^2$

Schrodinger equation for the rigid rotator $-\frac{\hbar^2}{8\pi^2 \mu} \nabla^2 \psi = E \psi$

The equation in terms of Spherical polar co-ordinate is

$$-\frac{\hbar^2}{8\pi^2 \mu} \left[\frac{1}{r^2} \frac{\partial}{\partial r} \left(r^2 \frac{\partial}{\partial r} \right) + \frac{1}{r^2 \sin \theta} \frac{\partial}{\partial \theta} \left(\sin \theta \frac{\partial}{\partial \theta} \right) + \frac{1}{r^2 \sin^2 \theta} \frac{\partial^2}{\partial \phi^2} \right] \psi = E \psi$$

For the rigid rotator r is constant. So, $\frac{\partial \psi}{\partial r} = 0$

Therefore, $-\frac{\hbar^2}{8\pi^2 \mu} \left[\frac{1}{r^2 \sin \theta} \frac{\partial}{\partial \theta} \left(\sin \theta \frac{\partial}{\partial \theta} \right) + \frac{1}{r^2 \sin^2 \theta} \frac{\partial^2}{\partial \phi^2} \right] \psi = E \psi$

Where, $\psi = f(\theta, \phi)$ Let $\psi(\theta, \phi) = \Theta(\theta)\Phi(\phi)$

So, the equation become $\frac{\Phi}{r^2 \sin \theta} \frac{\partial}{\partial \theta} \left(\sin \theta \frac{\partial}{\partial \theta} \right) \Theta + \frac{\Theta}{r^2 \sin^2 \theta} \frac{\partial^2 \Phi}{\partial \phi^2} + \frac{8\pi^2 \mu}{\hbar^2} E \Theta \Phi = 0$

Dividing both sides by $\Theta \Phi$ we get

$$\frac{1}{r^2 \sin \theta} \frac{\partial}{\partial \theta} \left(\sin \theta \frac{\partial}{\partial \theta} \right) \Theta + \frac{1}{\Phi r^2 \sin^2 \theta} \frac{\partial^2 \Phi}{\partial \phi^2} + \frac{8\pi^2 \mu}{\hbar^2} E = 0$$

Multiplying both sides by $r^2 \sin^2 \theta$

$$\frac{\sin \theta}{\Theta} \frac{\partial}{\partial \theta} \left(\sin \theta \frac{\partial}{\partial \theta} \right) \Theta + \frac{1}{\Phi} \frac{\partial^2 \Phi}{\partial \phi^2} + \frac{8\pi^2 \mu}{\hbar^2} r^2 \sin^2 \theta E = 0$$

Or, $-\frac{1}{\Phi} \frac{\partial^2 \Phi}{\partial \phi^2} = \frac{\sin \theta}{\Theta} \frac{\partial}{\partial \theta} \left(\sin \theta \frac{\partial}{\partial \theta} \right) \Theta + \frac{8\pi^2 \mu r^2 E \sin^2 \theta}{\hbar^2}$

Consider $-\frac{1}{\Phi} \frac{\partial^2 \Phi}{\partial \phi^2} = \frac{\sin \theta}{\Theta} \frac{\partial}{\partial \theta} \left(\sin \theta \frac{\partial}{\partial \theta} \right) \Theta + \frac{8\pi^2 \mu r^2 E \sin^2 \theta}{\hbar^2} = m_l^2$

$$\text{So, } \frac{1}{\Phi} \frac{\partial^2 \Phi}{\partial \varphi^2} = -m_l^2$$

$$\text{Solving we get } \Phi = \frac{1}{\sqrt{2\pi}} e^{im_l \varphi}$$

$$\text{Again } \frac{\sin \theta}{\Theta} \frac{\partial}{\partial \theta} \left(\sin \theta \frac{\partial}{\partial \theta} \right) \Theta + \frac{1}{\Phi} \frac{\partial^2 \Phi}{\partial \varphi^2} + \lambda \sin^2 \theta = m_l^2 \quad \left[\lambda = \frac{8\pi^2 \mu r^2 E}{h^2} = \frac{8\pi^2 I E}{h^2} \right]$$

Its solution is complicated.

$$\text{Solving we get } E = l(l+1) \frac{h^2}{8\pi^2 I}$$

Angular momentum component operator

The angular momentum operator L_z in spherical polar coordinate

$$\hat{L}_z = -i\hbar \frac{\partial}{\partial \varphi} = \frac{h}{2\pi i} \frac{\partial}{\partial \varphi}$$

$$\text{Therefore, } \hat{L}_z \Phi = \frac{h}{2\pi i} \frac{\partial}{\partial \varphi} \left(\frac{1}{\sqrt{2\pi}} e^{im_l \varphi} \right) = \frac{h}{2\pi i} (im_l) \left(\frac{1}{\sqrt{2\pi}} e^{im_l \varphi} \right) = m_l \frac{h}{2\pi} \Phi$$

$$\text{Eigenvalue} = m_l \frac{h}{2\pi} \quad m_l = 0, \pm 1, \pm 2$$

$$\text{Classically energy of rotating particle } E = \frac{1}{2} \frac{L^2}{I}$$

$$\text{Quantum mechanically energy of rotating particle } E = l(l+1) \frac{h^2}{8\pi^2 I}$$

$$\text{So, } L^2 = l(l+1) \frac{h^2}{4\pi^2}$$

$$\text{Therefore, the magnitude of the } \textit{angular momentum} \quad L = \sqrt{l(l+1)} \frac{h}{2\pi}$$

The normalised wave function $\psi(\theta, \varphi) = \Theta(\theta)\Phi(\varphi)$ is denoted by $Y_{l,m_l}(\theta, \varphi)$. The normalised wave function $Y_{l,m_l}(\theta, \varphi)$ is known as *spherical harmonics*.

The first few spherical harmonics

Azimuthal quantum number, l	Magnetic quantum number, m_l	Spherical harmonics, $Y_{l,m_l}(\theta, \varphi)$
0	0	$\left(\frac{1}{4\pi}\right)^{1/2}$
1	0	$\left(\frac{3}{4\pi}\right)^{1/2} \cos \theta$
	± 1	$\mp \left(\frac{3}{8\pi}\right)^{1/2} \sin \theta e^{\pm i\varphi}$
2	0	$\left(\frac{5}{16\pi}\right)^{1/2} (3\cos^2 \theta - 1)$
	± 1	$\mp \left(\frac{15}{8\pi}\right)^{1/2} \cos \theta \sin \theta e^{\pm i\varphi}$

	± 2	$\left(\frac{15}{32\pi}\right)^{1/2} \sin^2 \theta e^{\pm 2i\phi}$
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Qualitative treatment of hydrogen atom and hydrogen like ions: Setting up of Schrodinger equation in spherical polar coordinates, radial part, quantization of energy (only final energy expression); Average and most probable distance of electron from nucleus; setting up of Schrodinger equation for many electron atoms (He, Li)

Hydrogen like system

Hamiltonian operator $\hat{H} = \hat{T} + \hat{V}$

For hydrogen like system $V(r) = -\frac{Ze^2}{(4\pi\epsilon_0)r}$ and $\hat{T} = -\frac{h^2}{8\pi^2\mu}\nabla^2$

Schrodinger equation for the rigid rotator $\hat{H}\psi = E\psi$

$$\text{i.e. } -\left[\frac{h^2}{8\pi^2\mu}\nabla^2 + \frac{Ze^2}{(4\pi\epsilon_0)r}\right]\psi = E\psi$$

The equation in terms of Spherical polar co-ordinate is

$$-\frac{h^2}{8\pi^2\mu}\left[\frac{1}{r^2}\frac{\partial}{\partial r}\left(r^2\frac{\partial}{\partial r}\right) + \frac{1}{r^2\sin\theta}\frac{\partial}{\partial\theta}\left(\sin\theta\frac{\partial}{\partial\theta}\right) + \frac{1}{r^2\sin^2\theta}\frac{\partial^2}{\partial\phi^2}\right]\psi(r,\theta,\phi) - \frac{Ze^2}{(4\pi\epsilon_0)r}\psi(r,\theta,\phi) - E\psi(r,\theta,\phi) = 0$$

Consider $\psi(r,\theta,\phi) = R(r)Y(\theta,\phi)$

$$\text{So, } -\frac{h^2}{8\pi^2\mu}\left[\frac{1}{r^2}\frac{\partial}{\partial r}\left(r^2\frac{\partial}{\partial r}\right) + \frac{1}{r^2\sin\theta}\frac{\partial}{\partial\theta}\left(\sin\theta\frac{\partial}{\partial\theta}\right) + \frac{1}{r^2\sin^2\theta}\frac{\partial^2}{\partial\phi^2}\right]R(r)Y(\theta,\phi) - \frac{Ze^2}{(4\pi\epsilon_0)r}R(r)Y(\theta,\phi) - ER(r)Y(\theta,\phi) = 0$$

Multiplying by r^2 we get

$$-\frac{h^2}{8\pi^2\mu}\left[\frac{1}{\sin\theta}\frac{\partial}{\partial\theta}\left(\sin\theta\frac{\partial}{\partial\theta}\right) + \frac{1}{\sin^2\theta}\frac{\partial^2}{\partial\phi^2}\right]R(r)Y(\theta,\phi) = \left[\frac{h^2}{8\pi^2\mu}\left\{\frac{\partial}{\partial r}\left(r^2\frac{\partial}{\partial r}\right)\right\} - \frac{Ze^2r}{(4\pi\epsilon_0)} + r^2E\right]R(r)Y(\theta,\phi)$$

$$\text{Or, } -R\left[\frac{1}{\sin\theta}\frac{\partial}{\partial\theta}\left(\sin\theta\frac{\partial}{\partial\theta}\right) + \frac{1}{\sin^2\theta}\frac{\partial^2}{\partial\phi^2}\right]Y = Y\frac{\partial}{\partial r}\left(r^2\frac{\partial}{\partial r}\right)R + \frac{8\pi^2\mu}{h^2}\left[\frac{Ze^2r}{(4\pi\epsilon_0)} + r^2E\right]RY$$

Dividing this by RY

$$-\frac{1}{Y}\left[\frac{1}{\sin\theta}\frac{\partial}{\partial\theta}\left(\sin\theta\frac{\partial}{\partial\theta}\right) + \frac{1}{\sin^2\theta}\frac{\partial^2}{\partial\phi^2}\right]Y = \frac{1}{R}\frac{\partial}{\partial r}\left(r^2\frac{\partial}{\partial r}\right)R + \frac{8\pi^2\mu}{h^2}\left[\frac{Ze^2r}{(4\pi\epsilon_0)} + r^2E\right]$$

$$\text{Consider } -\frac{1}{Y}\left[\frac{1}{\sin\theta}\frac{\partial}{\partial\theta}\left(\sin\theta\frac{\partial}{\partial\theta}\right) + \frac{1}{\sin^2\theta}\frac{\partial^2}{\partial\phi^2}\right]Y$$

$$= \frac{1}{R} \frac{\partial}{\partial r} \left(r^2 \frac{\partial}{\partial r} \right) R + \frac{8\pi^2 \mu}{h^2} \left[\frac{Ze^2 r}{(4\pi\epsilon_0)} + r^2 E \right] = l(l+1)$$

Angular equation $\left[\frac{1}{\sin\theta} \frac{\partial}{\partial\theta} \left(\sin\theta \frac{\partial}{\partial\theta} \right) + \frac{1}{\sin^2\theta} \frac{\partial^2}{\partial\phi^2} \right] Y(\theta, \phi) = -l(l+1)Y(\theta, \phi)$

Radial equation $\frac{\partial}{\partial r} \left(r^2 \frac{\partial}{\partial r} \right) R(r) + \frac{8\pi^2 \mu}{h^2} \left[\frac{Ze^2 r}{(4\pi\epsilon_0)} + r^2 E \right] = l(l+1)R(r) = l(l+1)$

Simplifying the above equations the energy eigen value

$$E = -\frac{2\pi^2 Z^2 \mu e^4}{(n^2 h^2)(4\pi\epsilon_0)^2} \quad n = 1, 2, 3 \dots$$

And $\psi_{nlm_l}(r, \theta, \phi) = R_{nl}(r)Y_{l,m_l}(\theta, \phi)$

Values of l	0	1	2	3	4
Notation of wave function	s	p	d	f	g

Hydrogenic Radial wavefunction $R_{nl}(r)$

Orbital	n	l	$R_{nl}(r)$
1s	1	0	$2 \left(\frac{Z}{a_0} \right)^{3/2} e^{-Zr/a_0}$
2s	2	0	$\frac{1}{\sqrt{2}} \left(\frac{Z}{a_0} \right)^{3/2} \left(1 - \frac{Zr}{2a_0} \right) e^{-Zr/2a_0}$
2p	2	1	$\frac{1}{2\sqrt{6}} \left(\frac{Z}{a_0} \right)^{5/2} r e^{-Zr/2a_0}$
3s	3	0	$\frac{2}{3\sqrt{3}} \left(\frac{Z}{a_0} \right)^{3/2} \left(1 - \frac{2Zr}{3a_0} + \frac{2Z^2 r^2}{27a_0^2} \right) e^{-Zr/3a_0}$
3p	3	1	$\frac{8}{27\sqrt{6}} \left(\frac{Z}{a_0} \right)^{3/2} \left(\frac{Zr}{a_0} - \frac{Z^2 r^2}{6a_0^2} \right) e^{-Zr/3a_0}$
3d	3	2	$\frac{1}{81\sqrt{30}} \left(\frac{Z}{a_0} \right)^{7/2} r^2 e^{-Zr/3a_0}$

Many electron atoms (He, Li)

Hamiltonian operator $\hat{H} = \hat{T} + \hat{V}$

(i) Kinetic energy operator for a n -dynamical identical particles $\hat{T} = -\frac{h^2}{8\pi^2 m} \sum_{i=1}^n \nabla_i^2$

for a n -dynamical non-identical particles $\hat{T} = -\frac{h^2}{8\pi^2} \sum_{i=1}^n \frac{1}{m_i} \nabla_i^2$

For helium atom $\hat{T} = -\frac{h^2}{8\pi^2 m_e} \nabla_1^2 - \frac{h^2}{8\pi^2 m_e} \nabla_2^2 = -\frac{h^2}{8\pi^2 m_e} (\nabla_1^2 + \nabla_2^2)$

For potential energy operator have two contributions, attraction potential energy and

repulsion potential energy $V = -\frac{Ze^2}{(4\pi\epsilon_0)r_1} - \frac{Ze^2}{(4\pi\epsilon_0)r_2} + \frac{e^2}{(4\pi\epsilon_0)r_{12}} = \frac{1}{4\pi\epsilon_0} \left[-\frac{Ze^2}{r_1} - \frac{Ze^2}{r_2} + \frac{Ze^2}{r_{12}} \right]$

$$\therefore \hat{H} = -\frac{\hbar^2}{8\pi^2 m_e} (\nabla_1 + \nabla_2) - \frac{1}{4\pi\epsilon_0} \left(\frac{Ze^2}{r_1} + \frac{Ze^2}{r_2} - \frac{Ze^2}{r_{12}} \right)$$

Schrodinger equation $\hat{H}\psi = E\psi$

Q. Calculate the expectation value of r for an electron in the ground state of hydrogen atom.

Ans. The normalized wave function for the ground state of hydrogen atom is

$$\begin{aligned} \psi_{1,0,0} &= R_{1,0}\Theta_{0,0}\Phi_0 = \left(\frac{1}{\pi a_0^3}\right)^{3/2} e^{-r/a_0} \\ \langle r \rangle &= \langle \psi_{1,0,0} | r | \psi_{1,0,0} \rangle = \int \left[\left(\frac{1}{\pi a_0^3}\right)^{3/2} e^{-r/a_0} r \left(\frac{1}{\pi a_0^3}\right)^{3/2} e^{-r/a_0} \right] d\tau \\ &= \int \left[\left(\frac{1}{\pi a_0^3}\right)^{3/2} e^{-r/a_0} r \left(\frac{1}{\pi a_0^3}\right)^{3/2} e^{-r/a_0} \right] d\tau \\ &= \left(\frac{1}{\pi a_0^3}\right)^3 \int_0^\infty r^3 e^{-2r/a_0} dr \int_0^\pi \sin\theta d\theta \int_0^{2\pi} d\phi \quad [d\tau = r^2 dr \sin\theta d\theta d\phi] \\ &= \left(\frac{1}{\pi a_0^3}\right)^3 \int_0^\infty r^3 e^{-2r/a_0} dr \int_0^\pi \sin\theta d\theta \int_0^{2\pi} d\phi \\ &= \left(\frac{1}{\pi a_0^3}\right)^3 \frac{3!}{(2/a_0)^4} (2)(2\pi) = \frac{3}{2} a_0 \end{aligned}$$

Q. Calculate the average distance of 2s electron from nucleus of H-atom.

Ans. $\psi_{2s} = \psi_{2,0,0} = R_{2,0}\Theta_{0,0}\Phi_0 = \left[\left(\frac{1}{2a_0}\right)^{3/2} \left(2 - \frac{r}{a_0}\right) e^{-r/2a_0}\right] \left(\frac{1}{\sqrt{2}}\right) \left(\frac{1}{\sqrt{2\pi}}\right)$

$$\begin{aligned} &= \left(\frac{1}{\sqrt{32\pi}}\right) \left(\frac{1}{a_0}\right)^{3/2} \left(2 - \frac{r}{a_0}\right) e^{-r/2a_0} \\ \langle r \rangle &= \langle \psi_{2,0,0} | r | \psi_{2,0,0} \rangle \\ &= \int \left[\left(\frac{1}{\sqrt{32\pi}}\right) \left(\frac{1}{a_0}\right)^{3/2} \left(2 - \frac{r}{a_0}\right) e^{-r/2a_0} r \left(\frac{1}{\sqrt{32\pi}}\right) \left(\frac{1}{a_0}\right)^{3/2} \left(2 - \frac{r}{a_0}\right) e^{-r/2a_0} \right] d\tau \\ &= \left(\frac{1}{32\pi}\right) \left(\frac{1}{a_0^3}\right) \int r^3 \left(2 - \frac{r}{a_0}\right)^2 e^{-r/a_0} dr \int_0^\pi \sin\theta d\theta \int_0^{2\pi} d\phi \\ &= \left(\frac{1}{32\pi}\right) \left(\frac{1}{a_0^3}\right) \int \left(4r^3 + \frac{r^5}{a_0^2} - \frac{4r^4}{a_0}\right) e^{-r/a_0} dr \int_0^\pi \sin\theta d\theta \int_0^{2\pi} d\phi \\ &= \left(\frac{1}{32\pi}\right) \left(\frac{1}{a_0^3}\right) \left[4 \frac{3!}{(1/a_0)^4} + \frac{1}{a_0^2} \frac{5!}{(1/a_0)^6} - \frac{1}{a_0} \frac{4!}{(1/a_0)^5} \right] (2)(2\pi) \\ &= 6 a_0 \end{aligned}$$

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Q. Calculate the most probable distance of electron in 1s orbital of H-atom.

Ans. $\frac{d}{dr}(r^2 R_{1,0}^2) = 0$

So, $\frac{d}{dr}(r^2 e^{-2r/a_0}) = 0$

Or, $\left[2r + r^2 \left(-\frac{2}{a_0}\right)\right] e^{-2r/a_0} = 0$

Or, $r = a_0$

Q. Calculate the most probable distance of electron in 2s orbital of H-atom.

Ans. $\frac{d}{dr}(r^2 R_{2,0}^2) = 0$

So, $\frac{d}{dr} \left[r^2 \left(2 - \frac{r}{a_0}\right)^2 e^{-r/a_0} \right] = 0$

Or, $\frac{d}{dr} \left[\left(4r^2 + \frac{r^4}{a_0^2} - \frac{4r^3}{a_0}\right) e^{-r/a_0} \right] = 0$

Or, $\left[\left(8r + \frac{4r^3}{a_0^2} - \frac{12r^2}{a_0}\right) + \left(4r^2 + \frac{r^4}{a_0^2} - \frac{4r^3}{a_0}\right) \left(-\frac{1}{a_0}\right) \right] e^{-r/a_0} = 0$

Or, $8r + \frac{8r^3}{a_0^2} - \frac{16r^2}{a_0} - \frac{r^4}{a_0^3} = 0$

Or, $\left(\frac{r}{a_0}\right)^3 - 8\left(\frac{r}{a_0}\right)^2 + 16\left(\frac{r}{a_0}\right) - 8 = 0$

Or, $\left[\left(\frac{r}{a_0}\right)^2 - 6\left(\frac{r}{a_0}\right) + 4\right] \left(\frac{r}{a_0} - 2\right) = 0$

$\therefore \frac{r}{a_0} - 2$ or, $r = 2a_0$

Again $\frac{r}{a_0} = \frac{6 \pm \sqrt{36 - 16}}{2} = 3 \pm \sqrt{5}$

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Radial distribution function $F_R = (\text{volume of spherical shell}) (\text{probability density})$
 $= (4\pi r^2 dr)(R^2)$

The radial distribution function means the total probability of finding the electron in a spherical shell of thickness dr located at the distance r from the nucleus.

Note: The charge within the shell (r and $r + dr$) is

$$\int_r^{r+dr} R^2 r^2 dr \int_0^\pi \Theta^2 \sin \theta d\theta \int_0^{2\pi} \Phi^2 d\phi$$

For normalized Θ and Φ functions

$$\int_r^{r+dr} R^2 r^2 dr \int_0^\pi \Theta^2 \sin \theta d\theta \int_0^{2\pi} \Phi^2 d\phi = \int_r^{r+dr} R^2 r^2 dr$$

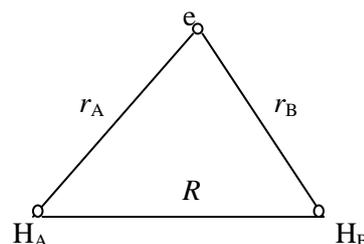
LCAO and HF-SCF: Covalent bond, valence bond and molecular orbital approaches, LCAO-MO treatment of H_2^+ ; bonding and antibonding orbitals; Qualitative extension to H_2 ; Comparison of LCAO-MO and VB treatment of H_2 and their limitations; Hartree-Fock method development, SCF and configuration interaction (only basics)

LCAO-MO of H_2^+

$$\psi_{MO} = C_1 \psi_{1s(A)} + C_2 \psi_{1s(B)}$$

The Schrodinger equation $H_{op} \psi_{MO} = E \psi_{MO}$

$$H_{op} = -\frac{\hbar^2}{8\pi^2 m} \nabla^2 - \frac{e^2}{(4\pi\epsilon_0)r_A} - \frac{e^2}{(4\pi\epsilon_0)r_B} + \frac{e^2}{(4\pi\epsilon_0)R}$$



$$E = \frac{\langle \psi_{MO} | H_{op} | \psi_{MO} \rangle}{\langle \psi_{MO} | \psi_{MO} \rangle} = \frac{\langle C_1 \psi_{1s(A)} + C_2 \psi_{1s(B)} | H_{op} | C_1 \psi_{1s(A)} + C_2 \psi_{1s(B)} \rangle}{\langle C_1 \psi_{1s(A)} + C_2 \psi_{1s(B)} | C_1 \psi_{1s(A)} + C_2 \psi_{1s(B)} \rangle}$$

$$= \frac{C_1^2 \langle \psi_{1s(A)} | H_{op} | \psi_{1s(A)} \rangle + C_2^2 \langle \psi_{1s(B)} | H_{op} | \psi_{1s(B)} \rangle + C_1 C_2 \langle \psi_{1s(A)} | H_{op} | \psi_{1s(B)} \rangle + C_1 C_2 \langle \psi_{1s(B)} | H_{op} | \psi_{1s(A)} \rangle}{C_1^2 \langle \psi_{1s(A)} | \psi_{1s(A)} \rangle + C_2^2 \langle \psi_{1s(B)} | \psi_{1s(B)} \rangle + 2C_1 C_2 \langle \psi_{1s(A)} | \psi_{1s(B)} \rangle}$$

$$= \frac{C_1^2 \langle \psi_{1s(A)} | H_{op} | \psi_{1s(A)} \rangle + C_2^2 \langle \psi_{1s(B)} | H_{op} | \psi_{1s(B)} \rangle + 2C_1 C_2 \langle \psi_{1s(A)} | H_{op} | \psi_{1s(B)} \rangle}{C_1^2 \langle \psi_{1s(A)} | \psi_{1s(A)} \rangle + C_2^2 \langle \psi_{1s(B)} | \psi_{1s(B)} \rangle + 2C_1 C_2 \langle \psi_{1s(A)} | \psi_{1s(B)} \rangle}$$

$$= \frac{C_1^2 \alpha_A + C_2^2 \alpha_B + 2C_1 C_2 \beta_{AB}}{C_1^2 + C_2^2 + 2C_1 C_2 S_{AB}} \quad [\beta_{AB}, \text{ resonance integral; } S_{AB}, \text{ overlapping integral}]$$

$$\text{Or, } E(C_1^2 + C_2^2 + 2C_1 C_2 S_{AB}) = C_1^2 \alpha_A + C_2^2 \alpha_B + 2C_1 C_2 \beta_{AB} \quad \dots (1)$$

Differentiating w.r.t C_1 we get

$$(\partial E / \partial C_1)(C_1^2 + C_2^2 + 2C_1 C_2 S_{AB}) + E(2C_1 + 2C_2 S_{AB}) = 2C_1 \alpha_A + 2C_2 \beta_{AB}$$

For at minima $\partial E / \partial C_1 = 0$

$$\therefore E(2C_1 + 2C_2 S_{AB}) = 2C_1 \alpha_A + 2C_2 \beta_{AB}$$

$$\text{Or, } C_1(\alpha_A - E) + C_2(\beta_{AB} - ES_{AB}) = 0$$

Similarly for $\partial E / \partial C_2$ we get

$$C_1(\beta_{AB} - ES_{AB}) + C_2(\alpha_B - E) = 0$$

In matrix form

$$\begin{vmatrix} \alpha_A - E & \beta_{AB} - ES_{AB} \\ \beta_{AB} - ES_{AB} & \alpha_B - E \end{vmatrix} \begin{vmatrix} C_1 \\ C_2 \end{vmatrix} = 0$$

For nontrivial solution determinant of coefficient is zero.

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$$\begin{vmatrix} \alpha_A - E & \beta_{AB} - ES_{AB} \\ \beta_{AB} - ES_{AB} & \alpha_B - E \end{vmatrix} = 0$$

Here, $\alpha_A = \alpha_B = \alpha$

$$\text{So, } (\beta_{AB} - ES_{AB})^2 - (\alpha - E)^2 = 0$$

$$\text{Or, } \alpha - E = \pm(\beta_{AB} - ES_{AB})$$

$$\text{So, the two roots are } E_+ = \frac{\alpha + \beta_{AB}}{1 + S_{AB}} \quad \text{and} \quad E_- = \frac{\alpha - \beta_{AB}}{1 - S_{AB}}$$

Using the above equation we get $C_1 = \pm C_2$

$$\psi_{+\text{MO}} = C_+(\psi_{1s(A)} + \psi_{1s(B)}) \quad \text{and} \quad \psi_{-\text{MO}} = C_-(\psi_{1s(A)} - \psi_{1s(B)})$$

$$\langle \psi_{+\text{MO}} | \psi_{+\text{MO}} \rangle = 1$$

$$\text{i.e. } \langle C_+(\psi_{1s(A)} + \psi_{1s(B)}) | C_+(\psi_{1s(A)} + \psi_{1s(B)}) \rangle = 1$$

$$\text{Or, } C_+^2 \{ \langle \psi_{1s(A)} | \psi_{1s(A)} \rangle + \langle \psi_{1s(B)} | \psi_{1s(B)} \rangle + 2 \langle \psi_{1s(A)} | \psi_{1s(B)} \rangle \} = 1$$

$$\text{Or, } C_+^2 (1 + 1 + 2S_{AB}) = 1$$

$$\therefore C_+ = \frac{1}{\sqrt{2(1 + S_{AB})}}$$

$$\text{Similarly } C_- = \frac{1}{\sqrt{2(1 - S_{AB})}}$$

$$\text{Now } \alpha_A = \langle \psi_{1s(A)} | H_{\text{op}} | \psi_{1s(A)} \rangle \quad \text{and}$$

$$H_{\text{op}} = -\frac{\hbar^2}{8\pi^2 m} \nabla^2 - \frac{e^2}{(4\pi\epsilon_0)r_A} - \frac{e^2}{(4\pi\epsilon_0)r_B} + \frac{e^2}{(4\pi\epsilon_0)R}$$

$$\begin{aligned} \text{So, } \alpha_A &= \langle \psi_{1s(A)} | -\frac{\hbar^2}{8\pi^2 m} \nabla^2 - \frac{e^2}{(4\pi\epsilon_0)r_A} - \frac{e^2}{(4\pi\epsilon_0)r_B} + \frac{e^2}{(4\pi\epsilon_0)R} | \psi_{1s(A)} \rangle \\ &= \langle \psi_{1s(A)} | -\frac{\hbar^2}{8\pi^2 m} \nabla^2 - \frac{e^2}{(4\pi\epsilon_0)r_A} | \psi_{1s(A)} \rangle + \langle \psi_{1s(A)} | -\frac{e^2}{(4\pi\epsilon_0)r_B} | \psi_{1s(A)} \rangle + \\ &\quad \langle \psi_{1s(A)} | \frac{e^2}{(4\pi\epsilon_0)R} | \psi_{1s(A)} \rangle \\ &= E_{1s(H)} + J + \frac{e^2}{(4\pi\epsilon_0)R} \end{aligned}$$

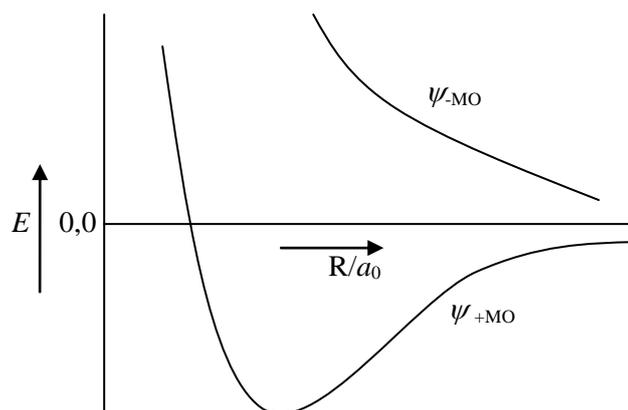
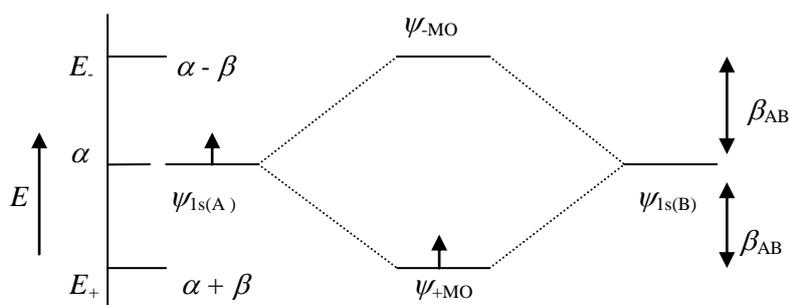
[J = Coulomb integral, electrostatic attraction between proton B and an electron in a 1s orbital centred on proton A]

$$\begin{aligned} \beta_{AB} &= \langle \psi_{1s(A)} | -\frac{\hbar^2}{8\pi^2 m} \nabla^2 - \frac{e^2}{(4\pi\epsilon_0)r_A} - \frac{e^2}{(4\pi\epsilon_0)r_B} + \frac{e^2}{(4\pi\epsilon_0)R} | \psi_{1s(B)} \rangle \\ &= \langle \psi_{1s(A)} | -\frac{\hbar^2}{8\pi^2 m} \nabla^2 - \frac{e^2}{(4\pi\epsilon_0)r_B} | \psi_{1s(B)} \rangle + \langle \psi_{1s(A)} | -\frac{e^2}{(4\pi\epsilon_0)r_A} | \psi_{1s(B)} \rangle + \\ &\quad \langle \psi_{1s(A)} | \frac{e^2}{(4\pi\epsilon_0)R} | \psi_{1s(B)} \rangle \end{aligned}$$

$$= E_{1s(H)}S_{AB} + K + \left(\frac{e^2}{(4\pi\epsilon_0)R} \right) S_{AB}$$

Now the exchange integral $K = \langle \psi_{1s(A)} | -\frac{e^2}{(4\pi\epsilon_0)r_A} | \psi_{1s(B)} \rangle$

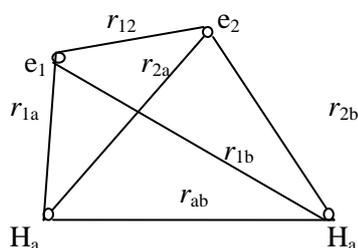
[Exchange of electron between A and B]



Hydrogen molecule H_2

MO

$$H_{op} = -\frac{\hbar^2}{8\pi^2m} \nabla_1^2 - \frac{\hbar^2}{8\pi^2m} \nabla_2^2 - \frac{e^2}{(4\pi\epsilon_0)r_{1a}} - \frac{e^2}{(4\pi\epsilon_0)r_{2a}} - \frac{e^2}{(4\pi\epsilon_0)r_{1b}} - \frac{e^2}{(4\pi\epsilon_0)r_{2b}} + \frac{e^2}{(4\pi\epsilon_0)r_{ab}} + \frac{e^2}{(4\pi\epsilon_0)r_{12}}$$



$$\psi = \psi_{\text{spatial}}\psi_{\text{spin}}$$

$$\psi_{\text{spatial}} = \psi_+(1)\psi_+(2)$$

$$\psi_{\text{spin}} = \frac{1}{\sqrt{2}}\{\alpha(1)\beta(2) - \alpha(2)\beta(1)\}$$

$$\text{In LCAO-MO} \quad \psi_+ = C_1\psi_{1s(\text{H}_a)} + C_2\psi_{1s(\text{H}_b)}$$

$$C_1 = C_2 = \frac{1}{\sqrt{2(1+S)}}$$

$$E = \frac{\langle \psi_+(1)\psi_+(2) | H_{\text{op}} | \psi_+(1)\psi_+(2) \rangle}{\langle \psi_+(1)\psi_+(2) | \psi_+(1)\psi_+(2) \rangle}$$

Calculated value -30.68 eV ($-2.96 \text{ MJ mol}^{-1}$) at internuclear distance 73 pm

Observed value -31.95 eV ($-3.08 \text{ MJ mol}^{-1}$) at internuclear distance 74.1 pm

VB

$$\psi = \psi_{1s(\text{H}_a)}(1)\psi_{1s(\text{H}_b)}(2)$$

$$H_{\text{op}} = H_a + H_b + H'$$

$$H_a = -\frac{\hbar^2}{8\pi^2m}\nabla_1^2 - \frac{e^2}{(4\pi\epsilon_0)r_{1a}} = -\frac{\hbar^2}{8\pi^2m}\nabla_2^2 - \frac{e^2}{(4\pi\epsilon_0)r_{2a}}$$

$$H_b = -\frac{\hbar^2}{8\pi^2m}\nabla_2^2 - \frac{e^2}{(4\pi\epsilon_0)r_{2b}} = -\frac{\hbar^2}{8\pi^2m}\nabla_1^2 - \frac{e^2}{(4\pi\epsilon_0)r_{1b}}$$

$$\begin{aligned} H' &= -\frac{e^2}{(4\pi\epsilon_0)r_{2a}} - \frac{e^2}{(4\pi\epsilon_0)r_{1b}} + \frac{e^2}{(4\pi\epsilon_0)r_{12}} + \frac{e^2}{(4\pi\epsilon_0)r_{ab}} \\ &= -\frac{e^2}{(4\pi\epsilon_0)r_{1a}} - \frac{e^2}{(4\pi\epsilon_0)r_{2b}} + \frac{e^2}{(4\pi\epsilon_0)r_{12}} + \frac{e^2}{(4\pi\epsilon_0)r_{ab}} \end{aligned}$$

$$E = \langle \psi | H | \psi \rangle$$

$$E = \langle \psi_{1s(\text{H}_a)}(1)\psi_{1s(\text{H}_b)}(2) | H_a + H_b + H' | \psi_{1s(\text{H}_a)}(1)\psi_{1s(\text{H}_b)}(2) \rangle$$

$$= \langle \psi_{1s(\text{H}_a)}(1) | H_a | \psi_{1s(\text{H}_a)}(1) \rangle \langle \psi_{1s(\text{H}_b)}(2) | \psi_{1s(\text{H}_b)}(2) \rangle$$

$$+ \langle \psi_{1s(\text{H}_b)}(2) | H_b | \psi_{1s(\text{H}_b)}(2) \rangle \langle \psi_{1s(\text{H}_a)}(1) | \psi_{1s(\text{H}_a)}(1) \rangle$$

$$+ E = \langle \psi_{1s(\text{H}_a)}(1)\psi_{1s(\text{H}_b)}(2) | H' | \psi_{1s(\text{H}_a)}(1)\psi_{1s(\text{H}_b)}(2) \rangle$$

$$E = E_{1s(\text{H}_a)} + E_{1s(\text{H}_b)} + Q$$

$$E_{\text{binding}} = E(H_2) - 2E(H) = Q$$

At 90 pm (r_{ab}) binding energy = 24 kJ mol^{-1}

Heitler and London wave function

$$\psi_{\text{VB}} = C_1\psi_{1s(\text{H}_a)}(1)\psi_{1s(\text{H}_b)}(2) + C_2\psi_{1s(\text{H}_a)}(2)\psi_{1s(\text{H}_b)}(1)$$

$$\psi_{\text{VB}} = C_1\psi_1 + C_2\psi_2$$

Where, $\psi_1 = \psi_{1s(\text{H}_a)}(1)\psi_{1s(\text{H}_b)}(2)$ and $\psi_2 = \psi_{1s(\text{H}_a)}(2)\psi_{1s(\text{H}_b)}(1)$

$$E = \frac{\langle \psi_{VB} | H_{op} | \psi_{VB} \rangle}{\langle \psi_{VB} | \psi_{VB} \rangle}$$

Minimizing E w.r.t C_1 and C_2 ($\partial E / \partial C_1 = 0$, $\partial E / \partial C_2 = 0$), we get

In matrix form

$$\begin{vmatrix} \alpha_1 - E & \beta_{12} - ES^2 \\ \beta_{12} - ES^2 & \alpha_1 - E \end{vmatrix} \begin{vmatrix} C_1 \\ C_2 \end{vmatrix} = 0$$

For nontrivial solution determinant of coefficient is zero.

$$\begin{vmatrix} \alpha_1 - E & \beta_{12} - ES^2 \\ \beta_{12} - ES^2 & \alpha_1 - E \end{vmatrix} = 0$$

Now $\alpha_1 = \alpha_2$

$$\text{So, } (\alpha_1 - E)^2 - (\beta_{12} - ES^2)^2 = 0$$

$$\text{Therefore, } E_+ = \frac{\alpha + \beta}{1 + S^2} \quad \text{and} \quad E_- = \frac{\alpha + \beta}{1 - S^2}$$

$$\text{So, } C_1 = \pm C_2 \quad C_1 = C_2$$

Many Electron Atoms

Hamiltonian operator for many electron system

$$H_{op} = -\frac{h^2}{8\pi^2 m} \sum_{i=1}^n \nabla_i^2 - \sum_{i=1}^n \frac{Ze^2}{(4\pi\epsilon_0)r_i} + \sum_{i=1}^n \sum_{j=i+1}^n \frac{e^2}{(4\pi\epsilon_0)r_{ij}}$$

Schrodinger equation

$$H_{op}\psi = E\psi$$

The Perturbation Method

In perturbation method electronic repulsion terms are consider as small perturbation on Hamiltonian operator

$$H_{op}^0 = -\frac{h^2}{8\pi^2 m} \sum_{i=1}^n \nabla_i^2 - \sum_{i=1}^n \frac{Ze^2}{(4\pi\epsilon_0)r_i}$$

$$H_{op} = H_{op}^0 + \sum_{i=1}^n \sum_{j=i+1}^n \frac{e^2}{(4\pi\epsilon_0)r_{ij}}$$

$$E = \frac{\int \psi^{0*} H_{op} \psi^0 d\tau}{\int \psi^{0*} \psi^0 d\tau}$$

Note: ψ^0 is not eigenfunction of H_{op} . So, not obtain exact energy eigenvalue.

$$\tilde{E} = \frac{\int \psi^{0*} H_{op}^0 \psi^0 d\tau + \int \psi^{0*} \left(\sum_{i=1}^n \sum_{j=i+1}^n \frac{e^2}{(4\pi\epsilon_0)r_{ij}} \right) \psi^0 d\tau}{\int \psi^{0*} \psi^0 d\tau}$$

$$\tilde{E} = E^0 + \frac{\int \psi^{0*} \left(\sum_{i=1}^n \sum_{j=i+1}^n \frac{e^2}{(4\pi\epsilon_0)r_{ij}} \right) \psi^0 d\tau}{\int \psi^{0*} \psi^0 d\tau}$$

For He atom, $Z = 2$. The second term is $\frac{5e^2}{(4\pi\epsilon_0)(4a_0)} = -\frac{5}{2}E_H = -108.8 \text{ eV}$

Where $E_H = -\frac{5e^2}{(4\pi\epsilon_0)(4a_0)} = -13.6 \text{ eV}$

Now $E^0 = E_1^0 + E_2^0 = \frac{Z^2}{n_1^2}E_H + \frac{Z^2}{n_2^2}E_H = 8E_H = 8(-13.6 \text{ eV}) = -108.8 \text{ eV}$

Where, $Z = 2$, $n_1 = n_2 = 1$

$\therefore E = -108.8 \text{ eV} + 34.0 \text{ eV} = -74.8 \text{ eV}$

The experimental value is -79.0 eV .

The Variational method

In variational method wavefunction is written with few adjustable parameters.

The choice of best orbital is **Slater orbital**. In Slater orbital angular part is same as Hydrogen like orbital but radial part is different.

$$R_{slater} = \left(\frac{2\zeta}{a_0}\right)^{n+1/2} \left[\frac{1}{(2n)!}\right] r^{n-1} e^{-\zeta r/a_0}$$

The parameter ζ is called orbital exponent.

$$\tilde{E} = \frac{\int \psi^* H_{op} \psi d\tau}{\int \psi^* \psi d\tau}$$

ψ is the approximate wave function.

$$\text{Now } \tilde{E} \geq E^0$$

Where, E^0 is the true value.

Minimizing the \tilde{E} with respect to adjustable parameter.

Consider He atom, in hydrogen like orbital Z is replace by Z' .

$$\psi = \left(\frac{1}{\pi}\right) \left(\frac{Z'}{a_0}\right)^3 \exp(-Z'r_1/a_0) \exp(-Z'r_2/a_0)$$

$$H_{op} = -\frac{\hbar^2}{8\pi^2 m} (\nabla_1^2 + \nabla_2^2) - \frac{Ze^2}{(4\pi\epsilon_0)r_1} - \frac{Ze^2}{(4\pi\epsilon_0)r_2} + \frac{e^2}{(4\pi\epsilon_0)r_{12}}$$

$$\text{and } \tilde{E} = \int \psi^* H_{op} \psi d\tau = \left[-2Z'^2 + \frac{27}{4}Z'\right] E_H$$

$$\text{Now } \left(\frac{\partial \tilde{E}}{\partial Z'}\right) = \left(-4Z' + \frac{27}{4}\right) E_H = 0$$

$$\therefore Z' = \frac{27}{16}$$

$$\left[-2\left(\frac{27}{16}\right)^2 + \left(\frac{27}{4}\right)\left(\frac{27}{16}\right)\right] (-13.6 \text{ eV}) = -77.5 \text{ eV}$$

SCF (Self-Consistent Field) Method (Hartree – 1928)

(i) Total wave function of a system is a product of one-electron wave function.

$$\psi = \varphi(1)\varphi(2)\dots\varphi(n)$$

One electron wave function is written as trial function.

(ii) Each electron is moving in an average spherically symmetric potential energy V_i . The potential energy arises due to nuclear attraction and repulsion from all other electrons. V_i obtain numerically using electrons wave function.

(iii) Write one electron Schrodinger equation

$$-\frac{h^2}{8\pi^2m}\nabla_i^2\phi_i + V_i\phi_i = E_i\phi_i \qquad \tilde{E} = \sum_{i=1}^n \tilde{E}_i$$

$$H_{\text{op}}\psi_{\text{MO}} = E\psi_{\text{MO}}$$

$$\text{Or,} \quad \left[2r + r^2 \left(-\frac{2}{a_0} \right) \right] e^{-2r/a_0} = 0$$

three(i) Vector addition $\vec{a} + \vec{b} = \vec{b} + \vec{a}$

$$(\vec{a} + \vec{b}) + \vec{c} = \vec{a} + (\vec{b} + \vec{c})$$

(ii) Multiplication of vector by a scalar $|\lambda\vec{a}| = |\lambda||\vec{a}|$

(iii) Components of vector $\vec{r} = x\hat{i} + y\hat{j} + z\hat{k}$