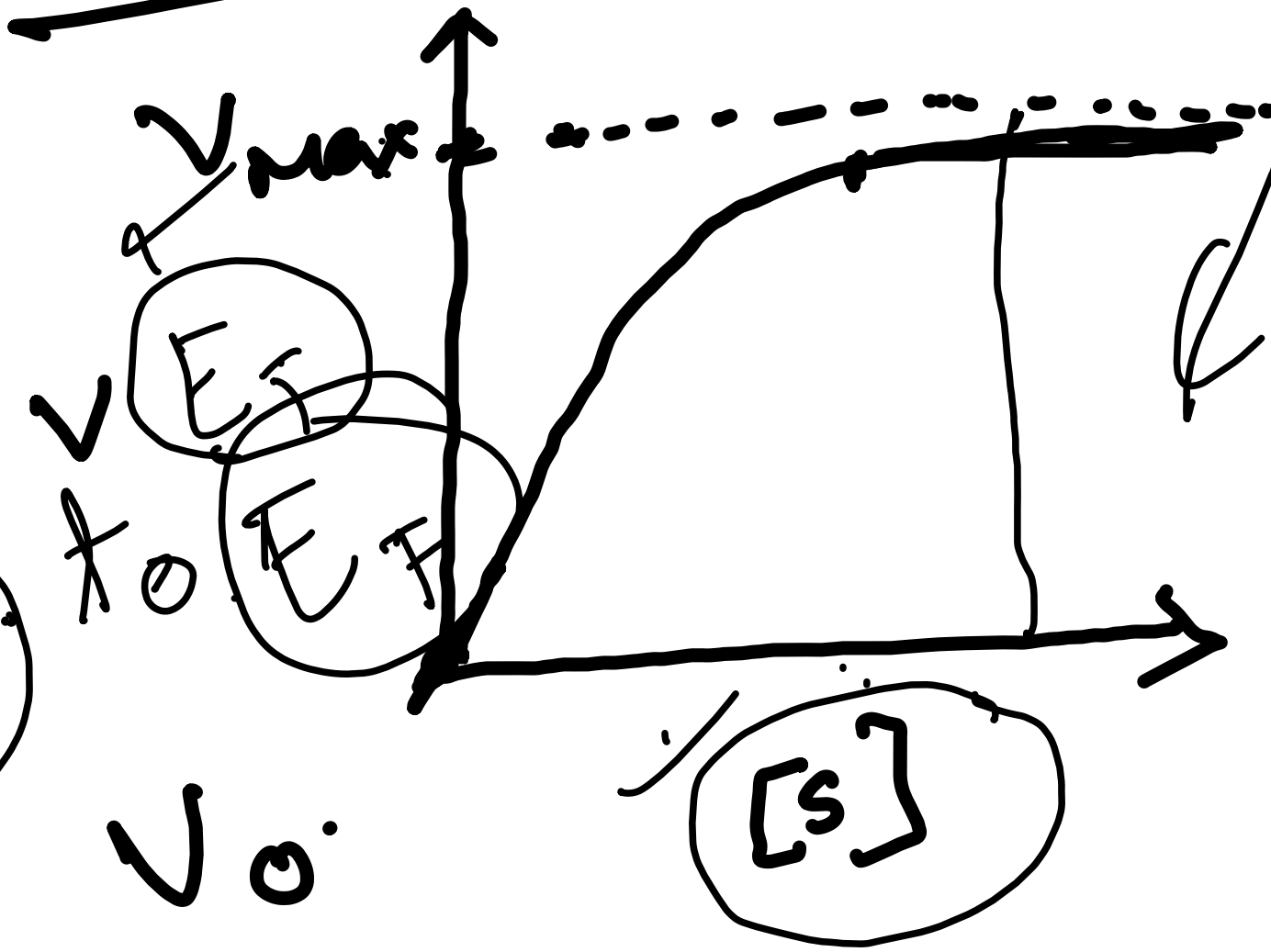


# Michaelis Menton Derivation

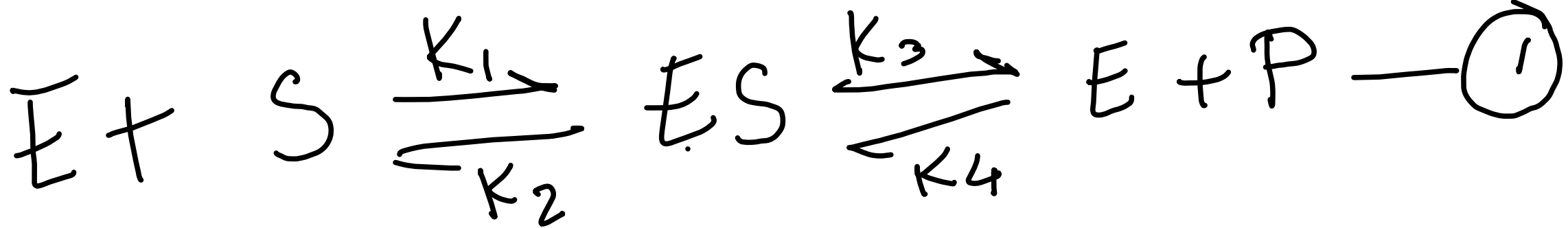
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# Enzymkinetik



aktive Zentren

• Co-ent  
• Enzymes  
• active sites  
• substrate  
• Enzyme → complex  
• Enzyme → product  
• Enzyme → product  
• Enzyme → product



$$V_0 = k_3 [ES] \quad \text{--- (2)}$$

$$[ES] = [E]_T \quad \text{at } V_{max}$$

$$V_{max} = k_3 [E]_T \quad \text{--- (3)}$$

At steady state  $\rightarrow$  equilibrium

• Rate of formation of ES =  $k_1 [E] [S]$

• Rate of consumption of ES =  $k_2 [ES] + k_3 [ES]$

Rate of consumption of ES = Rate of formation of ES

$$k_2 [ES] + k_3 [ES] = k_1 [E] [S] \quad \text{--- (4)}$$

$$\frac{k_2 + k_3}{k_1} = \frac{[E] [S]}{[ES]} \quad \text{--- (5)}$$

$$\frac{k_2 + k_3}{k_1}$$

$$K_M = \frac{k_2 + k_3}{k_1} \quad \text{--- (6)}$$

$K_M \rightarrow$  Michaelis constant.

- express  $[E]$  in terms of  $[ES]$  &  $[E]_T$

$$[E] = [E]_T - [ES] \quad \text{--- (7)}$$

- Subs (6) & (7) eq. (5)

$$K_M = \frac{([E]_T - [ES])[S]}{[ES]} \quad \text{--- (8)}$$

$$* K_M = \frac{[E_T] - [ES]}{[S]}$$

$$\Rightarrow [ES] K_M = [E_T][S] - [ES][S]$$

$$\Rightarrow [ES] K_M + [ES][S] = [E_T][S]$$

$$\Rightarrow [ES] (K_M + [S]) = [E_T][S]$$

$$\Rightarrow [ES] = \frac{[E_T][S]}{K_M + [S]}$$

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$$[ES] = \frac{[E]_T [S]}{K_M + [S]} \quad \dots \quad (9)$$

Subs (9) in (2)

$$V_0 = \frac{V_{max} [E]_T [S]}{K_M + [S]} \quad \dots \quad (10)$$

Subs eq (3) in eq (10)

$$V_0 = \frac{V_{max} [S]}{K_M + [S]}$$

→ MM eq